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¹H n.m.r. relaxation in bituminous coal and semicoke

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Temperature and frequency-dependent proton relaxation measurements, in conjunction with spin concentrations determined by e.s.r., have shown that for fresh and oxidized coal the relaxation mechanism involves spin-diffusion to paramagnetic centres and to rotating methyl groups on alkyl chains. ¹H relaxation in semicokes is more complex; probably both direct relaxation to paramagnetic centres as well as spin-diffusion contribute.

(Keywords: coal; semicoke; n.m.r.)

Recently there has been a great increase in interest in the n.m.r. properties of coal and related materials¹⁻⁵. ¹H n.m.r. relaxation times of different ranks of coal have been found to range over more than an order of magnitude^{1,2}. Although some attempts have been made to understand the relaxation mechanism, empirical relationships are frequently explored^{4,5}.

For high rank coals ($\approx 86-95 \text{ wt}\%$ carbon daf), T_1 increases with increasing radical content^{1,2} and it has been proposed that the relaxation mechanism involves spin-diffusion to paramagnetic centres¹. However, this does not explain the trend for lower rank coals, for which T_1 decreases with decreasing radical content. Also, possible contributions from other relaxation mechanisms have not been explored.

In order to elucidate ¹H relaxation mechanisms in coal further, the frequency and temperature dependences of the spin-lattice relaxation time of a high volatile bituminous coal have been examined in the present work, and the unpaired spin concentration has been determined. Fresh and oxidized samples of both coal and semicoke were studied.

EXPERIMENTAL

 T_1 was measured at 18, 34, 60 and 180 MHz on Bruker SXP and CXP pulsed n.m.r. spectrometers by means of $90^{\circ}-t-90^{\circ}$ pulse sequences. Temperature variation was accomplished in the cryostat described previously⁶. ¹H linewidths at room temperature were measured at 180 MHz from the Fourier transformed free induction decay. The analyses of the high volatile bituminous Devco coal used in these experiments are summarized in *Table 1*. The samples were sealed in glass tubes under vacuum after pumping for several hours. Spin concentrations were measured at 9.5 GHz on a Varian E-3 e.s.r. spectrometer using diphenyl picryl hydrazyl (DPPH) as a calibrant.

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Table 1 Analyses of coal samples

	Fresh	Oxidized			
		312 h, 100°C	192 h, 190° C		
Moisture (wt% ad)	0.8	0.9	0.4		
Ash (wt% ad)	2.4	2.3	2.1		
Volatile matter (wt% ad)	34.3	31.7	38.2		
Fixed carbon (wt% ad)	62.6	65.2	59.3		
Carbon (wt% ad)	84.3	82.1	67.8		
Hydrogen (wt% ad)	5.3	5.0	2.5		
Nitrogen (wt% ad)	2.1	2.2	1.8		
Sulphur (wt% ad)	0.7	0.6	0.6		
Oxygen (wt% ad) ^a	4.4	6.9	24.9		

a By difference

RESULTS AND DISCUSSION

For each coal sample, the magnetization decay curves were slightly non-exponential. The time constant governing the decay at long times was taken to be the characteristic T_1 value. Relaxation times at different frequencies are summarized in Table 2.

The temperature behaviour of T_1 for the coal samples is shown in Figure 1. For each sample there is a shallow minimum in T_1 , at 143 K for the 18 MHz measurements and at 167 K for the 60 MHz measurements.

Line shapes were approximately Gaussian and showed no evidence of two-component behaviour. The $\approx 8G$ linewidth leads to a second moment of $\approx 11G^2$, indicating that extensive molecular motion does not take place.

The relaxation behaviour of the fresh coal and lightly oxidized coal is much the same, T_1 values for the oxidized sample being slightly shorter at all temperatures (Figure 1). For bituminous coals containing ≈ 86 wt% carbon it has been suggested that relaxation is due to spin-diffusion to paramagnetic centres. Spin concentrations, as

Table 2 Summary of relaxation times and linewidths at room temperature

Sample	Linewidth (gauss)	T_1 (s)				No. of also
		18 MHz	34 MHz	60 MHz	180 MHz	No. of spins 10 ¹⁹ g ⁻¹
Coal						
Fresh	8.0	0.167	0.236	0.337	0.573	0.88
Oxidized 312 h, 100° C	8.1	0.134	0.187	0.219	0.486	1.5
Oxidized 192 h, 190° C	6.1	0.27	_	0.46	0.873	0.50
Semicoke						
Fresh	8.1	0.022	0.018	0.017	0.046	4.0
Oxidized	7.8	0.033	0.031	0.031	0.078	4.5

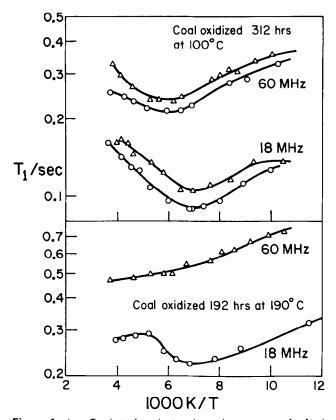


Figure 1 Log T_1 plotted against reciprocal temperature for fresh and oxidized coal samples at different frequencies. \triangle , Fresh coal; oxidized coal

determined by e.s.r. (Table 2), show that the unpaired-spin content increased on mild oxidation but decreased in the case of severe oxidation. The ¹H spin-lattice relaxation times at different frequencies decreased on mild oxidation and increased on severe oxidation, as shown in Table 2 and also in the log-log plot of T_1^4 against ω^2 (Figure 2). The fact that this plot gives reasonably straight lines for the three samples confirms that the main relaxation mechanism involves spin-diffusion to paramagnetic centres. Hence the proton T_1 values should vary inversely with unpaired spin content, as is indeed the case. Such a relaxation mechanism also explains the non-exponential initial magnetization decay⁸.

On the other hand, the minimum observed in the plot of $\log T_1$ against 1/T and the frequency dependence of the T_1 minimum is reminiscent of a classical BPP T_1 minimum due to molecular motion⁹. At the relatively low temperatures of the T_1 measurements, the only motion rapid enough to give a T_1 minimum in a complex material such as coal is methyl group rotation. The usual criteria for T_1 minima due to motions cannot be expected to apply, as methyl rotation is not the exclusive relaxation

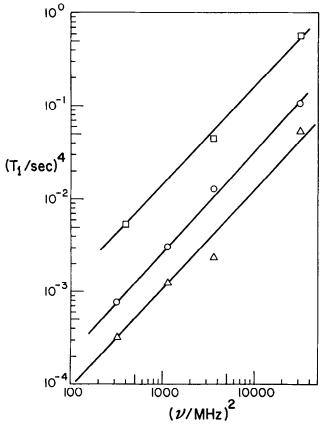


Figure 2 Log-log plot of the fourth power of relaxation time against frequency squared for fresh and oxidized coal samples. ○, Fresh coal; △, coal oxidized 312 h at 100°C; □, coal oxidized 192 h at 190°C

mechanism. Also, because of the non-crystalline nature of coal, there must be a distribution of motional parameters¹⁰, and the usual dependence of T_1 on the square of the frequency at low temperatures and the frequency independence of T_1 above the T_1 minimum can no longer hold 10 . However, at the T_1 minimum the relation $\omega \tau_{\rm c} \approx 0.62$ is valid and the frequency dependence of the T_1 minimum can be used to estimate an activation energy for the motion from

$$E_{\rm a} = \frac{R \ln(\omega_{\rm a}/\omega_{\rm b})}{1/T_{\rm b} - 1/T_{\rm a}} \tag{1}$$

where ω_a , ω_b = the radio frequencies; and T_a and T_b = the temperatures of the corresponding relaxation minima. The above equation applied to the data in Figure 1 yields an activation energy of 11.3 kJ mol⁻¹, a reasonable value for methyl groups terminating alkyl chains¹¹. Methyl groups directly attached to aromatic rings have much lower activation energies 12,13.

A rough estimate can be made of the number of methyl groups required to give a minimum of the observed depth. In the presence of spin-diffusion to methyl groups, T_1 is given by

$$T_1^{-1} = f_{Me} T_{1Me}^{-1} + (1 - f_{Me}) T_{1Me}^{-1}$$
 (2)

where f_{Me} = the fraction of protons in methyl groups; $T_{1 Me}$ is the relaxation time for an isolated methyl group; and $T_{\rm IN}$ is the relaxation time of the other protons, If we estimate $T_{\rm IN}$ to be 400 ms (Figure 1) and calculate the minimum $T_{1\text{Me}}$ to be 33 ms from the modified BPP equation¹⁴, f_{Me} is found to be 0.063. Because a distribution in motional parameters makes relaxation at the minimum less effective 10 , this value of f_{Me} represents a lower limit. For the heavily oxidized coal, the same T_1 minimum due to methyl group rotation is still evident, although at higher temperatures a new relaxation process sets in. This is in agreement with the fact that the coal structure is altered considerably during the oxidation process. The smaller proton linewidth as compared with the linewidth for the other two coal samples, as well as the appearance of a new relaxation mechanism, suggest increased molecular mobility, perhaps due to depolymerization.

Whereas the details of the variation in relaxation time with temperature apparently reflect molecular motion, the gross T₁ values reflect changes in radical content. Mild oxidation increases the radical content, but more severe oxidation causes a subsequent decrease. This is in agreement with a recent n.m.r. and e.s.r. study¹⁵ of the radical content as a function of oxidation for several coking coals.

As is evident in Table 2, the ¹H relaxation time in the semicokes is an order of magnitude shorter than in the coals. An attempt to measure T_1 at lower temperatures showed that the magnetization decay curves became extremely non-exponential. Since a lower proton density is to be expected as a consequence of the absence of volatiles, it is perhaps surprising that the proton linewidth is unaltered from that of fresh coal.

The frequency dependence of T_1 at room temperature for the semicokes does not show a simple trend and evidently does not fit a plot of T_1^4 against ω^2 . If extensive aromatization has taken place during the coking process, considerable delocalization of any unpaired spins present in the sample may be expected. Perhaps in addition to spin-diffusion to radical centres, many protons are relaxed directly by interactions with unpaired spins.

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