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# Characterization of ethylene-propylene-diene monomer (EPDM) roofing membranes using thermogravimetry and dynamic mechanical analysis

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#### Abstract

The effect of heat-aging at 100 °C and 130 °C on three EPDM roofing materials was studied by thermogravimetry (TGA), dynamic mechanical thermal analysis (DMA) and tensile testing. It was found that both TGA and DMA can be used in a routine manner to explain the changes in mechanical properties of EPDM roofing membranes. In general, the results showed that the degradation of the material increased with heating temperature and exposure duration. It was also shown that aging at elevated temperatures can reduce the time required to evaluate the effect of heat on mechanical properties of polymer-based roofing materials.

#### INTRODUCTION

Roofing membranes have been used for many years. In the past, the testing of various roofing materials was simple because of the accumulated experience and the lack of variety encountered. This, however, has changed radically with the availability of many different types of polymeric or polymer-based roofing membranes. Since these different types vary from modified-bitumen to PVC and EPDM-based roofing membranes in terms of chemical, physical, and mechanical characteristics, it is quite difficult to compare them solely on the basis of mechanical properties as has generally been done in the past. This is especially true since the mechanical properties of these materials are not only related to macroscopic behaviour but also at the molecular level which is controlled by chemical composition and molecular arrangement. Mechanical tests only describe how a material behaves with respect to a certain test and do not explain why the material failed unless the failure is related solely to structural properties. Thus, the mechanical tests cannot really assist in solving the problem and/or in the development of methods to assess the long-term behaviour of materials.

Thermal analysis is a technique that can provide some insight as to why some roofing materials fail more prematurely than others. Thermogravimetry (TGA) can be used to monitor changes in chemical composition of a material during weathering and/or exposure to elevated temperatures. Dynamic mechanical analysis (DMA or DMTA) can be used routinely to determine various properties of a roofing membrane, e.g. the glass transition temperature. Together, DMA and TGA can provide a measure of the degradative factors which can lead to the correlation between accelerated weathering and actual field performance.

In general, the roofing industry has used mainly mechanical tests for characterization and diagnostic work. The objective of this study is to demonstrate the utility of TGA and DMA in establishing the durability of roofing membranes because although these techniques have been applied to polymers [1] and other construction materials, application of these techniques is relatively new in the roofing field. An effort is made to correlate the mechanical properties of the roofing materials with thermal analyses and field service.

### EXPERIMENTAL

The thermogravimetric and dynamic mechanical analyses as well as the mechanical tests were performed on control and heat-aged EPDM roofing samples. A typical composition of an EPDM roofing membrane is given in Table 1.

# Heat-aging

Three commercially available, non-reinforced EPDM roofing membrane samples (S1, S2 and S3) were obtained. Each sample was cut into seven pieces of about  $150 \times 250$  mm of which one piece was used as a control and three were placed in air-circulating ovens preheated to  $100^{\circ}$ C and  $130^{\circ}$ C

#### TABLE 1

Typical composition of a generic EPDM roofing membrane

Ingredients	wt.% (approx.)	Function		
Synthetic rubber or blend	26	Imparts elastomeric properties		
Carbon black	37	Reinforces elastomeric properties and absorbs UV, acts as filler		
Extender oils	28	Aid processing and reduce cost		
Curing agents	4	Promotes proper cure		
Antioxidants and antiozonants	3	Assist in resisting environmental factors		
Processing aids	2	Promotes anti-sticking, and provides smooth surface in calendering		

for heat-aging. One piece of each sample was removed from the oven after 1, 7, and 28 days. This provided a total of 105 specimens for tensile testing and thermal analyses.

#### Tensile testing

The heat-aged material was conditioned at room temperature  $(22 \pm 2^{\circ} C)$  for a minimum of 24 h prior to testing. The specimens were cut from the control and aged samples, using a dumbbell-shaped die in a hydraulic press. The tests were carried out at room temperature using an Instron tensile tester (model 1122) with a gauge length of 60 mm and cross-head speed of 60 mm min<sup>-1</sup>. The load and elongation curves were recorded on the machine chart recorder as well as being stored in the computer. Five specimens were tested for each condition and the mean values of load and elongation-at-break were recorded. These are listed in Table 2.

### Thermogravimetric analysis

The weight-loss of each EPDM roofing sample was monitored by thermogravimetric analysis using a Du Pont Instruments model 951 interfaced to a 9900 and 1090 computer. A piece of approximately 25–40 mg of sample was cut using a hole puncher to ensure a uniform shape and better contact with the sample pan. The analyses were carried out at temperatures between  $20 \,^{\circ}$ C and  $800 \,^{\circ}$ C using a nitrogen gas flow of 100 ml min<sup>-1</sup> and a scanning rate of  $20 \,^{\circ}$ C min<sup>-1</sup>. Throughout this study the derivative TGA (DTG) curve was used as it provided the desired quantitative data. The data obtained from the DTG curves are not shown on the curves themselves but, for greater clarity, are tabulated in Table 3 as a function of heat-aging temperature and exposure-time.

Heat-aging schedule		Strength (kN m <sup>-1</sup> )			Elongation (%)		
Temp. (°C)	Days	<b>S</b> 1	S2	S3	<b>S</b> 1	S2	<b>S</b> 3
22 ª	0	10.6	8.7	11.4	522	285	307
100	1	10.9	8.8	10.8	433	266	314
100	7	11.6	9.5	11.3	396	211	297
100	28	10.7	10.5	10.6	324	132	270
130	1	9.9	9.2	9.3	348	180	273
130	7	9.7	10.9	9.8	307	88	227
130	28	8.0	11.6	9.7	218	38	159

# TABLE 2

Tensile strength and el	longation values	of EPDM	membranes
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<sup>4</sup> Unheated control material.

#### TABLE 3

Heat-aging schedule		Oil loss <sup>b</sup> (%)		Polymer loss <sup>c</sup> (%)			
Temp. (°C)	Days	<b>S</b> 1	S2	<b>S</b> 3	<b>S</b> 1	S2	<b>S</b> 3
22 <sup>a</sup>	0		20.0		52.5	22.5	52.5
100	1		13.2		54.0	25.5	51.8
100	7		7.2		53.4	27.7	51.9
100	28		4.9		54.5	29.1	51.6
130	1		5.1		54.0	28.5	52.8
130	7		2.2		53.6	30.0	52.4
130	28		1.6		51.6	31.2	51.8

TGA data for oil and polymer component content in the EPDM samples prior to and after heat-aging at 100  $^\circ C$  and 130  $^\circ C$ 

<sup>a</sup> Unheated control sample.

<sup>b</sup> Based on the integrated area of peak between 200 and 400 °C (not available for S1 or S3).

<sup>c</sup> Based on the integrated area of peak between 400 and 550 °C. In the case of S1 and S3, the area most probably includes both oil and polymer loss.

# Dynamic mechanical analysis

A Polymer Laboratory dynamic mechanical thermal analyser (PL-DMTA Mark I) was used with software version 4.0. The EPDM control and aged sheets were cut in the machine direction (1 mm thick, 5 mm long, 12 mm deep) and were run in the DMA in single cantilever mode using test frame L and clamp type C. A fixed frequency of 10 Hz and a heating rate of  $2^{\circ}$ C min<sup>-1</sup> was used to study the samples between -120 and  $+80^{\circ}$ C. In this system, a sinusoidal stress is applied to a vibrator unit using a sinusoidal current. The stress imposed upon the specimen is therefore directly related to the amount of current delivered to the vibrator. The sample is also subjected to strain via the induced displacement.

The glass transition temperature ( $\alpha$ -transition) values were obtained from the maximum in the loss modulus (E'') vs. temperature curve. The glass transition temperature data for all three samples are summarized in Table 4.

#### **RESULTS AND DISCUSSION**

The DTG curves for the EPDM sample S1 before and after heat-aging for 1, 7 and 28 days at 100 °C and 130 °C, are displayed in Fig. 1. It can be seen that in each case the elastomer degrades in the temperature range of about 300-550 °C. The very weak shoulder observed in the 400-450 °C range might be due to the volatilization of oils and/or plasticizers, which are typical ingredients in EPDM membranes (see Table 1). Since the volatilization of the oils and/or plasticizers overlap with the decomposition region of the polymer, percent oil weight-loss in the sample could not be determined. This type of thermal curve is frequently observed when the oils and/or

#### **TABLE 4**

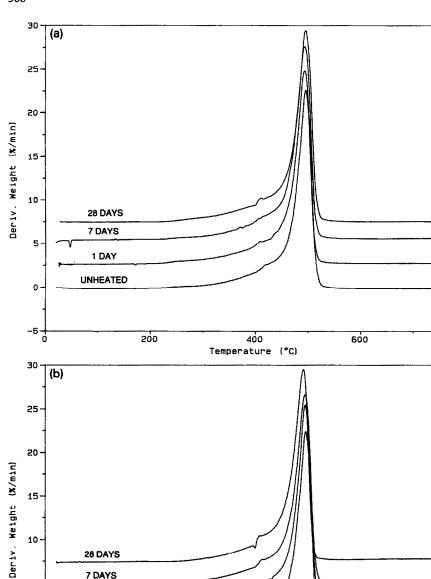
Heat-aging schedule		$T_{\rm g}$ (°C) for samples			
Temp. (°C)	Days	S1	S2	<b>S</b> 3	
22 ª	0	49	-61	- 46	
100	1	- 48	- 58	- 45	
100	7	- 49	- 50	- 42	
100	28	- 50	- 46	- 44	
130	1	- 48	- 50	- 46	
130	7	- 49	- 46	- 45	
130	28	- 48	-43	- 44	

Glass transition temperatures of EPDM membranes

<sup>a</sup> Unheated control material.

plasticizers are of low volatility [1]. From the curves, it can be seen that there is no significant degradation of the EPDM until approximately  $350 \,^{\circ}$ C, which is well above the heat-aging temperature. This may explain why, as will be shown later, no significant difference is observed in the strength and glass-transition temperatures between the unheated and the heat-aged specimens exposed for 1, 7 and 28 days at  $100 \,^{\circ}$ C and  $130 \,^{\circ}$ C (see Tables 2 and 4).

The derivative-TGA thermograms for the S2 specimens before and after heat-aging are given in Fig. 2. It can be clearly seen that the material degrades via a two-step process yielding weight-losses between 150 and 430°C, and between 440 and 550°C. The first weight-loss is most likely to arise from the volatilization of oils and/or plasticizers while the second weight-loss corresponds to the degradation of the polymer. Contrary to the S1 DTG curves (Fig. 1), the volatilization of oil does not overlap with the weight-loss region of the polymer, which may indicate that the oil in sample S2 is more volatile than that in S1. It is therefore possible to estimate the weight-loss for both the oil and polymer components of S2. As can be seen in Table 3, 20% of the unheated (control) material is volatilized in the 150-430°C range while 22.5% of the polymer component decomposes in the 440-550°C range. In the case of the samples heated at 100°C and 130°C, it was observed by TGA that the oil weight-loss decreased with time of exposure. For example, after 28 days of heat-aging at 130°C, 1.6% oil was detected (i.e. 92% of the oil content has been lost during heat-aging) by TGA vs. 5.1% (i.e. 75% of the oil was lost during the heat treatment) after 1 day of heat-aging. Furthermore, as the heat-aging temperature increases, the amount of oil lost decreases, i.e. after 1 day heat-aging at 130°C, the EPDM lost 5.1% of the oil (or 75% of the oil was volatilized during the heat-aging) compared to 13.2% (or 33% of the oil was lost in the heat-aging process) after 1 day heat-aging at 100°C. It was also noticed that the polymer loss increased as the oil loss decreased (Fig. 3) and physically, the sample became



800

800

Fig. 1. Derivative TGA (DTG) curve for the S1 EPDM sample heat-aged at (a) 100 °C and (b) 130°C, for 0 (unheated), 1, 7 and 28 days.

400

Temperature (°C)

600

10

5

0

-5+

28 DAYS 7 DAYS

1 DAY UNHEATED

200

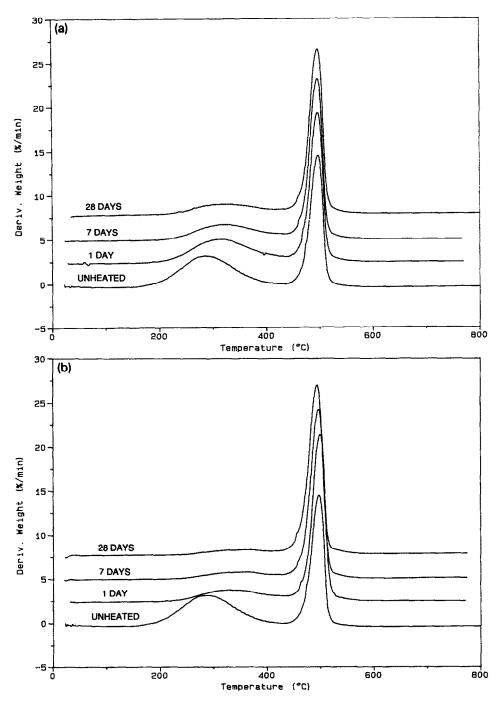


Fig. 2. Derivative TGA (DTG) curve for the S2 EPDM sample heat-aged at (a)  $100^{\circ}$ C and (b)  $130^{\circ}$ C, for 0 (unheated), 1, 7 and 28 days.

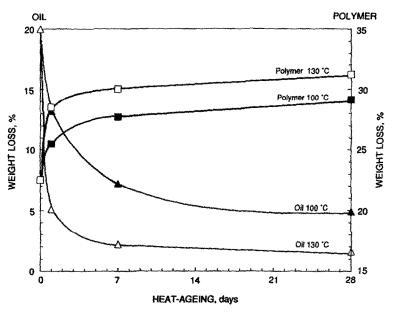


Fig. 3. Relationship between the weight-loss for the oil and polymer component for specimen S2.

brittle. This may indicate that the oils are used as plasticizers (e.g. hydrocarbon oils [2]) and/or stabilizers as well as processing agents, or that the weight-loss includes oils, stabilizers and plasticizers. All of the TGA data imply that during the heat treatment of S2, oil is volatilized and as a result the polymer component degrades more rapidly.

The DTG curves for the EPDM samples S3, (control and heat-aged) plotted in Fig. 4 show that, as in the case of S1, the degradation of the material occurs in one step. A shoulder appears in the 400-450 °C range, but it is more pronounced and better defined than for S1. As was previously mentioned, this shoulder may be attributed to volatilization of the oils and/or plasticizers.

Although all the samples have been heat-aged below the temperature where degradation begins (i.e. below 200 ° C), they still show some visual signs of degradation due to the length of exposure time. This is especially true for S2. The DTG curves for S1 (Fig. 1), S2 (Fig. 2) and S3 (Fig. 4) seem to indicate that the oils and/or plasticizers used in the S1 and S3 EPDM materials are of similar volatilities though the oil in S1 would appear to be more stable than that found in S3. Moreover, the oil for both S1 and S3 is considerably different from that used in sample S2, which is more volatile. In addition, some degree of stabilization against thermal degradation is possible through the use of additives [3,4]. Therefore, the different oil and/or the use of additives may explain the higher thermal stability of the S1 and S3 samples compared to that of S2.

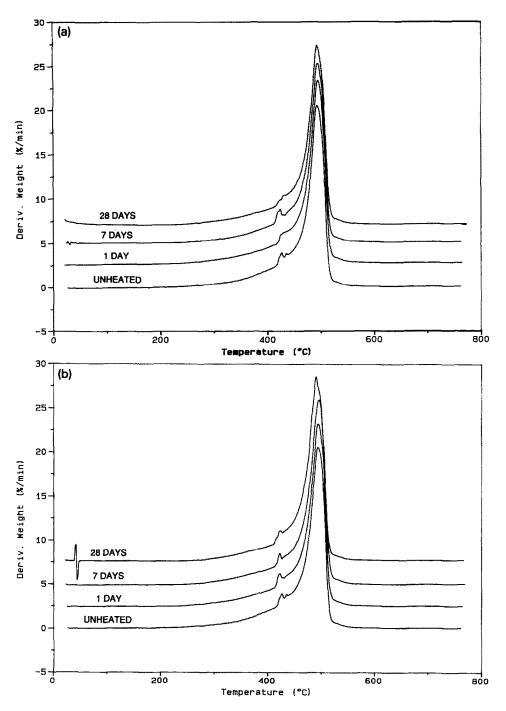


Fig. 4. Derivative TGA (DTG) curve for the S3 EPDM sample heat-aged at (a)  $100^{\circ}$ C and (b)  $130^{\circ}$ C, for 0 (unheated), 1, 7 and 28 days.

The use of oils and/or plasticizers in polymers is to improve the processability of the compounds, reduce cost, improve low temperature performance, etc. [1]. Hydrocarbon oils are often used because they can act as plasticizers and fillers [2]. As mentioned earlier, S2 loses the oils and/or plasticizers at much lower temperatures than do either S1 or S3. In addition, the percent weight-loss decreases with increasing temperature and time duration, which indicates that part of the volatile component may have been lost during the aging process. As a result, the sample loses some of its initial flexibility, inducing glass-like behaviour [3]. This will affect the tensile strength of the material and hence, the material will fail more easily than S1 and S3 EPDM. As can be seen from the tensile and elongation data (Table 2), the strength of S1 and S3 membranes remains fairly constant or decreases slightly. For S2, however, the strength at break increases with heat-aging, indicating that the sample is becoming more brittle. The elongation-at-break data for non heat-aged material is greater for both S1 and S3 than for S2. Perhaps this shows that the non-volatile oils are better plasticizers than the volatile ones and/or that there is a greater polymer component of S1 and S3 than S2. The elongation-at-break data, however, indicates that all three membrane samples have lost some of their elongation capabilities after heat-aging. This would signify that even when little or no oil is lost during the heat-aging process, the property of the membrane is changed. It is possible that, even if the processing oils are not completely lost during the heat treatment, their quality degrades and consequently they do not lubricate (i.e. plasticize) efficiently. If this were the case, then the elongation of the samples would decrease without seeing any significant change in glass transition temperature  $(T_g)$  of the roofing membrane. In order to verify this, the three roofing membranes were studied by DMA.

The dynamic mechanical analysis curves for sample S1, showing the loss modulus (E'') as a function of temperature can be found in Fig. 5 for heat-aging at 100°C and 130°C, respectively. Sample S1 undergoes very little change in glass transition temperature even after 28 days of heating at 100°C and 130°C. Sample S2, however, exhibits a significant change in  $T_g$ . Figures 6a (S2 heat-aged at 100°C) and 6b (S2 heat-aged at 130°C) show the DMA curves for sample S2 heat-aged for 0, 1, 7 and 28 days. It is apparent that  $T_g$  increased with the number of days of heating. For example, in the case of S2 at 100°C, the  $T_g$  values changed by approximately 20°C, i.e. from -61°C for no treatment to -46°C after 28 days of heat-aging, while at 130°C the glass transition temperature surged from -61°C to -50°C after only one day of accelerated aging. Following this sudden jump, the  $T_g$  gradually increased to -43°C after 28 days of heat-aging at 130°C. Sample S3 (Fig. 7) appears to behave in a similar fashion to S1, i.e. little change in glass transition temperature is observed after accelerated aging.

As can be seen from the  $T_g$  values shown in Table 4, only S2 (as predicted

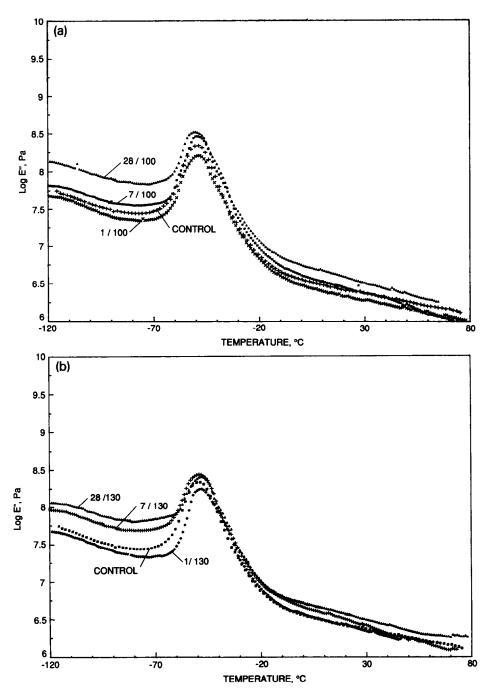
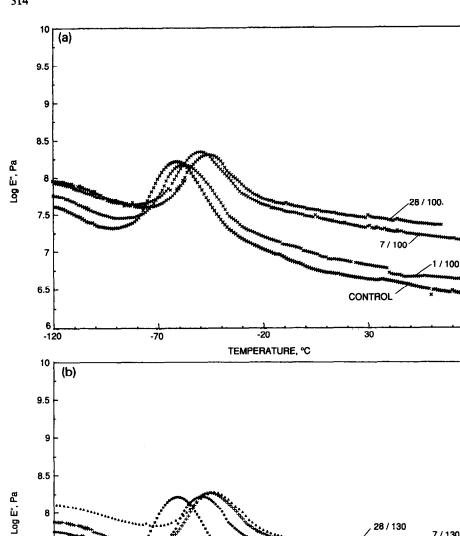


Fig. 5. DMA curves for specimen S1 aged at (a)  $100 \degree C$  for 0 days (control), 1 day (1/100), 7 days (7/100) and 28 days (28/100) and (b)  $130 \degree C$  for 0 days (control), 1 day (1/130), 7 days (7/130) and 28 days (28/130).



80

28/130

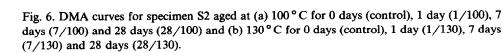
CONTROL

30

7/130

80

1/130



-20

TEMPERATURE, °C

-70

8

7.5

7

6.5

6

-120

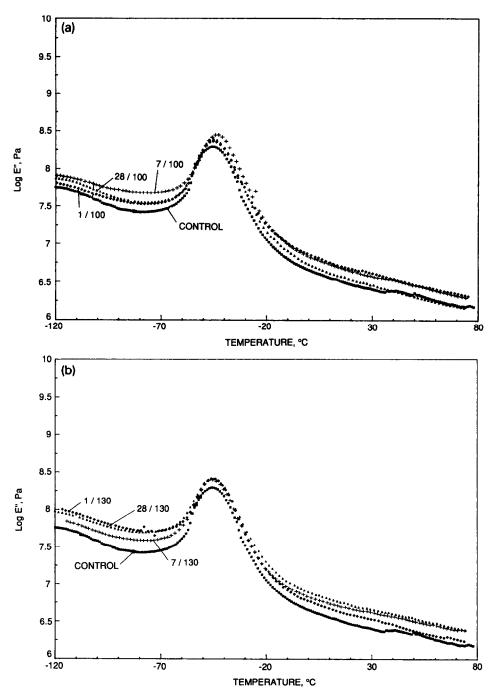


Fig. 7. DMA curves for specimen S3 aged at (a)  $100 \degree C$  for 0 days (control), 1 day (1/100), 7 days (7/100) and 28 days (28/100) and (b)  $130 \degree C$  for 0 days (control), 1 day (1/130), 7 days (7/130) and 28 days (28/130).

by the TGA data) is affected to any great extent by the heat-aging process. Although the  $T_g$  of S2 after 28 days of heat-aging is similar to S1 and S3, the fact that it is shifting to higher temperatures indicates that it is becoming more crosslinked and its physical/mechanical properties are changing accordingly (Table 2). As has been mentioned previously, all three roofing membranes sustained a change in their elongation-at-break capacity with this change being more dramatic for sample S2. Since the  $T_g$  did not vary to any great extent for samples S1 and S3, our earlier hypothesis that the quality of the oil is deteriorating could very well be true. The fact that the  $T_g$ values for S2 are changing quite extensively could be explained by an increase in crosslinking and/or the loss of plasticizers which might be contained in the oil (or the oil itself might be acting as a plasticizer).

It should be noted that S2 would have been considered the most appropriate membrane to use on roofs in extremely cold climates if no heat-aging had taken place. The importance of heat-aging is therefore quite obvious and must be carried out to properly evaluate a series of membranes.

It may be of interest to note the apparent equivalencies in the behaviour of materials aged at different temperatures. From the TGA and DMA data (Tables 3 and 4), it would appear that between 1 and 7 days of heat-aging at 130°C is equivalent to 28 days of heat-aging at 100°C. The equivalency is most obvious for the S2 sample where the percent weight-loss of oil is 4.9%, while that for the polymer component is 29.1% after 28 days at 100°C. compared with 5.1% and 28.5% for the oil and polymer components, respectively, after only one day at 130°C. The glass transition temperature for S2 is -46°C after 28 days at 100°C and 7 days at 130°C indicating apparent equivalency. This indicates that accelerated aging, induced by exposing membranes to elevated temperature, may reduce the time required at the maximum service temperature to evaluate the effect of heat on mechanical properties of polymer-based materials. It is much more difficult to make any statement with regard to equivalency for S1 and S3 since the observed changes in weight-loss and  $T_g$  are minimal and could easily be due to experimental error.

#### CONCLUSIONS

EPDM roofing materials were characterized using thermogravimetry, dynamic mechanical analysis and tensile testing. It was shown that both TGA and DMA can be used to evaluate the properties of various roofing membranes. The use of a dynamic analyser facilitates the determination of the glass transition temperature which can be used to explain the change in mechanical properties of roofing membranes. The volatilities and polymer components of a roofing membrane can be determined by TGA. These methods can be used to assess the validity of the accelerated aging processes that expose membranes to elevated temperatures, and thus reduce the time required to evaluate the effect of heat on the mechanical properties of polymer-based materials. Finally, the equivalency of material response at different temperatures could help in establishing standard requirements.

#### ACKNOWLEDGEMENTS

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