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Development of Gas Pressure Cells for Diffraction Experiments:

hurdles, headaches and practical considerations

Pamela Whitfield, James Ross, Lyndon Mitchell and Victoria Nawaby



National Research Council Canada Conseil national de recherches Canada



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Pressure cells at ICPET the beginning...

- Polymer researchers were interested in the crystallization behaviour of bio-polymers under sub- and supercritical CO₂.
- Decided to construct a cell for *in-situ* lab XRD
 - rated for 125 bar (1800psi) dry CO₂, 200°C (recirculating)
 - intended for use with MoK α (17.5 keV) and PSD detector
- Existing literature design re-visited with finite element analysis
 modifications made for compliance with the ASME pressure code
- 304SS pressure vessel (Nitronic 60 for other parts)
- 1/8" thick coated beryllium windows
- Elastomer seals (started with Viton but changed to Aflas)

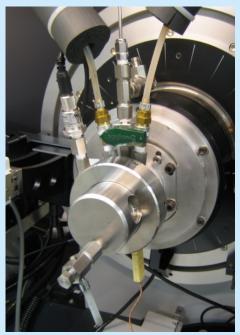
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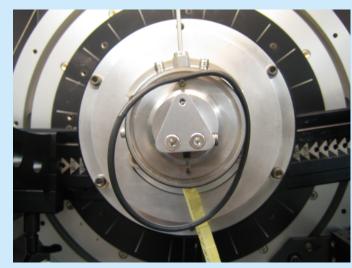
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Polymer crystallization under CO₂ pressure

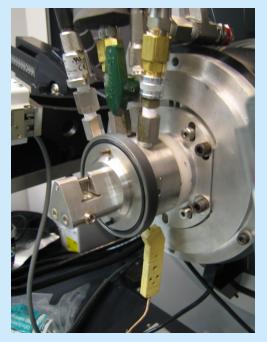
- The cell took ~2 years from concept to delivery
- The cell worked very well but shortly after commissioning the polymer researcher left NRC and the project ended.

The Gen1 stage mounted on a Bruker D8 diffractometer





Viton gets the 'bends' with high pressure CO_2 . Aflas resists explosive decompression



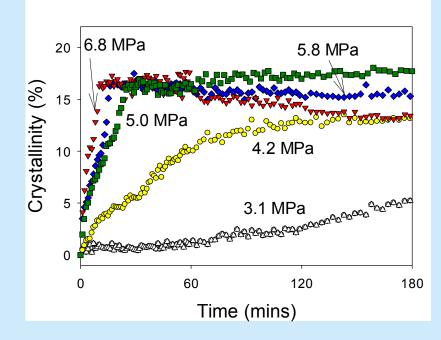
Cover held with a threaded collar

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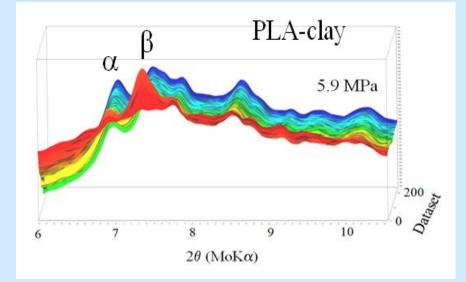
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Polylactic Acid

- Crystallization of PLA showed pressure dependence
- Phase changes with some samples...



Crystallization and phase evolution between α and β -PLA in a polylactic acid–clay composite under dry CO₂



Crystallization over a period of 3 hours of PLA-clay composites under dry CO_2



- Had unique expertise and equipment but needed a new direction....
- Was suggested that such technology could be useful for studies relating to reaction kinetics in CO₂ sequestration
 - geochemical models rely on reaction kinetics data to make long term predictions on the fate of CO₂ underground
 - however, there is a serious lack of experimental data on possible reaction kinetics under down-hole conditions
- The big question can a lab pressure cell give us anything useful?

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Proof of concept using Gen1 cell

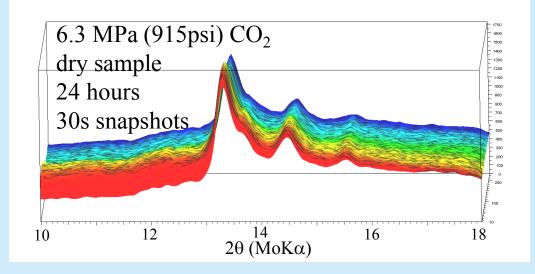
- Gen1 stage very limited
 - not designed for wet experiments have to improvise
 - 304SS susceptible to corrosion high purity H_2O needed
- Targets for a proof of concept
 - synthetic calcium silicate hydrate very reactive
 - wollastonite reactive and readily available in pure form
 - calcined lizardite more reactive than raw serpentine
- Samples had to be damp before heating
 - water is necessary for gas transport and reaction
 - limits upper temperature due to evaporation

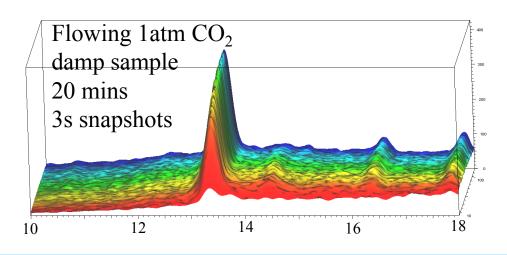
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- Calcium silicate hydrate (C-S-H) is the main binding phase of cement and very reactive with CO₂
- Without water no reaction at 915psi over a full day
- A couple of drops of water and reaction almost instantaneous with flowing CO₂
 - struggled to get good enough time resolution

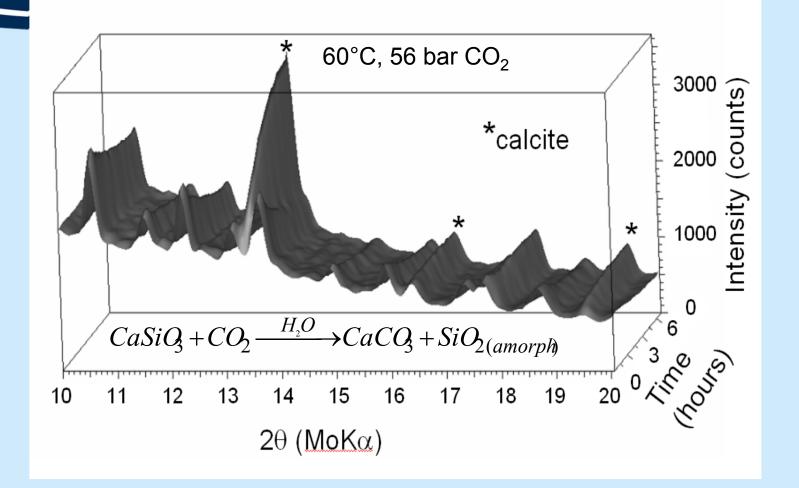
Effect of water – synthetic C-S-H carbonation





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Wollastonite – CaSiO₃

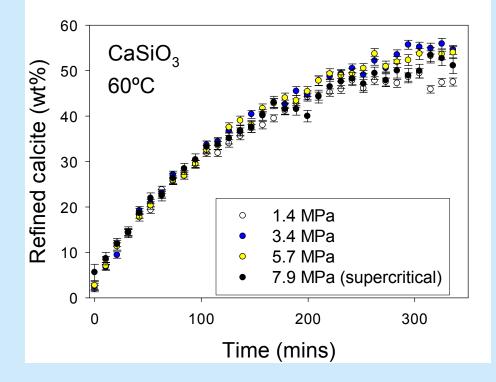


Time-resolved XRD data showing carbonation of damp wollastonite over 6 hours

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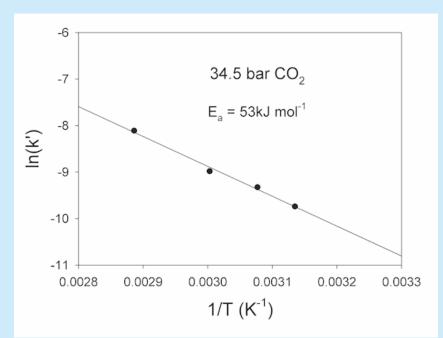
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Wollastonite – CaSiO₃



Arrhenius plot for carbonation of CaSiO₃ to give activation energy for reaction

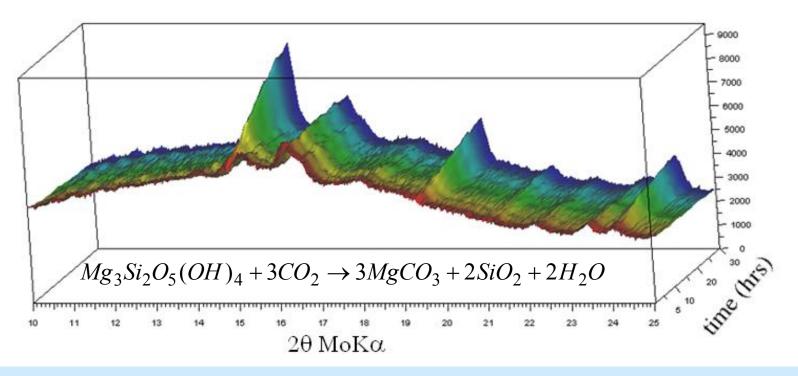
Carbonation at 60°C under different CO_2 pressures. Reaction rate not affected when CO_2 supercritical.



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Lizardite (the nonchrysotile serpentine)

 Mg silicates much more sluggish reactions – had to calcine it to get reaction under these conditions



Carbonation of damp, calcined lizardite over 30 hours (4 minute snapshots, 125 °C, 35 bar CO_2)

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The Next Step...

- A custom-designed cell to simulate down-hole conditions
- Would still have to meet CSA regulations and ASME pressure code but working conditions much more severe
 - 300°C max
 - 300 bar (4350 psi) max
 - concentrated brine (e.g. Salton Sea)
 - acidic conditions (pH3)
 - sour conditions
 - possible impurity gases (e.g. SO₂)
- CO₂, H₂S and CH₄ can be supercritical in this range of conditions

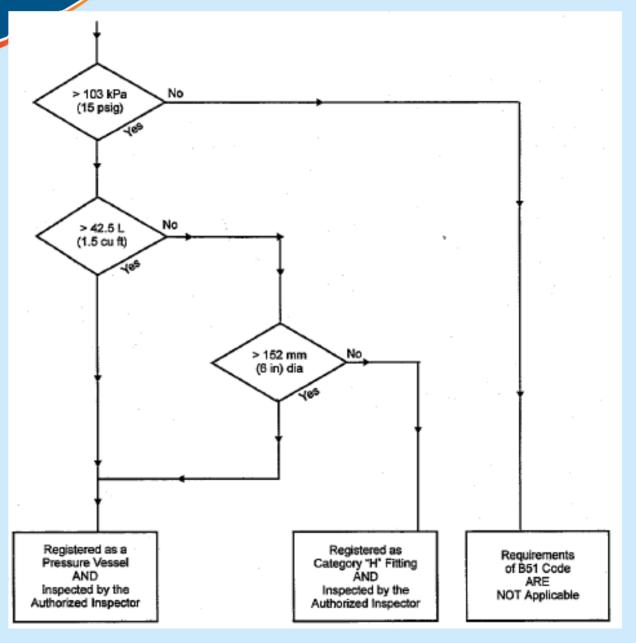
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Decision-tree for vessels containing lethal gases under CSA regulations

Due to small size and gas volume the cell classified as a fitting.

Note that in this tree car tires would be classed as pressure vessels! Tires are specifically exempted under the regulations....



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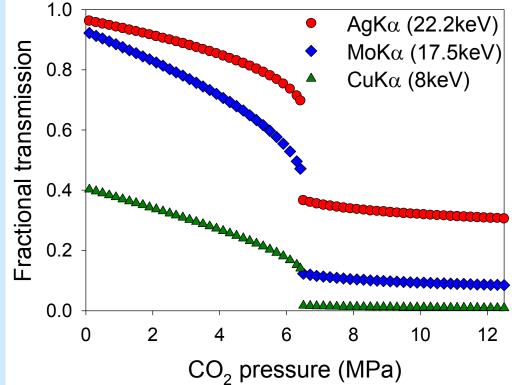
Lab & Synchrotron as Complimentary Sources

- High pressure experiments usually done at synchrotrons
 - brightness, tuneable wavelength, detector coverage, resolution...
- Very good for fast reactions
 - e.g. reactions with impurities in the CO₂ stream
- However.....
 - many reactions in proposed CO₂ reservoirs are very sluggish
 - getting kinetic information could take weeks or even longer....
 - synchrotron beamtime isn't available on that timescale
- Lab-based systems are less powerful, slower but....
 - if the reactions are slow then....
 - the X-ray source can be tailored to the problem

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Lab Systems - the energy problem...

- Lab systems are limited by the available X-ray tubes and generators
- Density of CO₂ can approach H₂O at high pressures
- Even MoKα won't punch through – too much attenuation
- Highest practical energy is a silver anode X-ray tube but very rare



Theoretical room temperature X-ray transmission (Gen1) at increasing CO_2 pressure with different X-ray tubes

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Challenges in Building the Gen2 Cell

- Have to satisfy pressure regulations and ASME code
 - ASME boiler and pressure vessel code written for industry
 - materials for a 10 ton boiler not exactly suitable
 - have to get it right first time, on-the-fly modifications not allowed



- Space we have to make the assembled stage small and light enough to fit on a lab diffractometer $\theta \theta$ goniometer
- Design for both reflection and transmission with same cell
- Have to find a TSSA-approved pressure-vessel company willing to work on an all-machined fitting
 - TSSA is the Ontario regulatory body for pressure vessels

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Materials Selection

- Materials have to withstand extremely corrosive conditions
 - candidates for vessel were alloys suitable for geothermal wells
 - NACE not enough, had to be ASME compliant
- Grade 20 titanium would be great but its not code-approved
- Settled on C-22 for best corrosion resistance rather than strength
 - bar-stock >4" diameter acceptable under Section VIII, division II ASME but not acceptable under B31.3 piping code
 - had to get a 4" forging made to be code compliant
 - lack of consistency in different sections so gets very complicated...



Nuclear



Sections VIII, X, XII

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- The windows were the biggest problem...
 - Beryllium very sensitive to chloride corrosion but can get certified stress-strain behaviour for structural grade Be
- Would be great to use thinner diamond of sapphire windows but
 - materials properties not certified so can't use them
 - tests to failure on windows not good enough...
- Brush-Wellman recommended Ta coating
 - AgK α can get through it
 - window assembly designed to avoid damaging this coating
 - each window ¼" thick and costs \$2500!
- Nickel alloy c-rings used for window and cover seals
 - The seal with c-rings actually improves with pressure

Notched to stop window ~ rotating when seal tightened

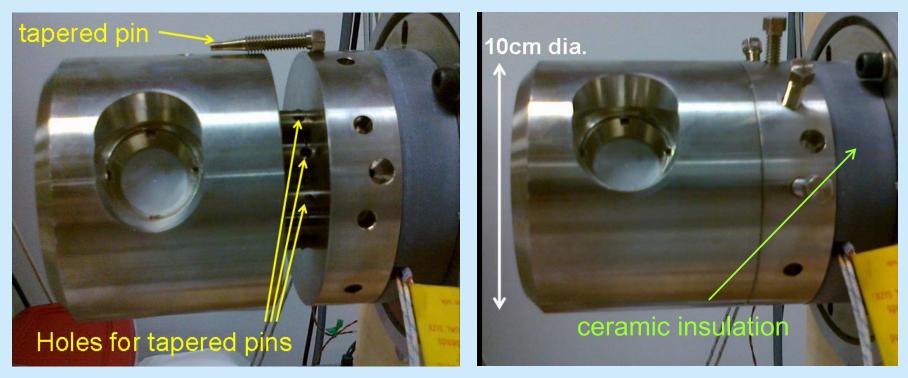
c-ring

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What it looks like...

- Complete redesign from the ground up
 - little in common with Gen1 design



Cover held by 12 tapered pins seating into retaining bolts. The Gen1 design using a locking collar warped during FEA analysis under the stress of Gen2 conditions.

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Other features

- Electrically heated
 - 3 sets independently controlled for a more even temperature (reduce stress)
- Multipurpose internal slot
 - integrated Ta knife-edge for reflection
 - sample holder for transmission experiments





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The design process

- Basic design done at the NRC design and Fabrication Services – including full FEA and thermal modelling
 - done to ASME section VIII division 2 (design by analysis)
 - certified as process piping under ASME B31.3 (small volume)
- A prototype was built to test thermal behaviour of the cell
 - checking for hot spots that would increase local stresses



Instrumented mock-up of the Gen2 stage made for testing thermal behaviour. Inset is a infra-red image of the stage at an internal temperature of 300°C.

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More Regulatory Headaches....

- As per Ontario regulations the design had to be fully reviewed, reanalyzed and signed off by professional engineers (in this case All-Weld)
- The design was then submitted to the Ontario regulator for yet another review (third time..) and final design registration
- Now All-Weld can go ahead and make one.....
- Before delivery has to be hydrostatically tested to 1.5x maximum working pressure – 450 bar (6500 psi)

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You have the cell... Now what?

- The cell has to be part of an integrated system to get results
- Siemens don't make silver anode X-ray tubes any more
 - special order (Bruker persuaded them to make a small batch)
- Standard high speed detectors for lab systems don't work with $AgK\alpha$ need good efficiency make most of the weak signal
 - special custom Si-strip detector required
 - for transmission work a custom $\mbox{AgK}\alpha$ focussing mirror is needed
- High pressure CO₂ syringe pump
- Corrosion-resistant valves/fittings

The most expensive Swagelok relief valve you'll ever see...

Ta-coated, special Kalrez o-ring for valve seating





- The whole process has taken over 2 years from design concept to finish......
- All-Weld due to complete construction by mid-March
- After testing delivery is expected end of March
- Commissioning and testing to follow on lab diffractometer

- Then we're open for business!
- Integration of cell on a NSLS beamline will follow....



- The whole process is one very big headache
- Pressure regulations make life very difficult and ironically can be less safe with necessary compromises on materials
- Canada has the toughest pressure vessel regulations in the world
 - the US is actually a bit easier...
- For CO₂ sequestration work both synchrotron and lab-based studies will play a role
 - some reactions are too slow for synchrotrons unless you have your own beamline to play with....

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Questions?

